Pseudouridine Transformations. Formation of 2'- and 3'-Deoxypseudouridines *via* Halogen Intermediates using α-Acetoxyisobutyryl Chloride

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Summary Treatment of pseudouridine (1) with α -acetoxyisobutyryl chloride (neat or in acetonitrile solution) at elevated temperatures followed by hydrogenolysis of the resulting mixture of chloro-sugar intermediates and deblocking gave the 2'- and 3'-deoxypseudouridines (2) and (3) (ca. 90%) in a ratio of ca. 55:45 with no evidence for the previously suggested α -anomer (4) of 2'-deoxypseudouridine.

PSEUDOURIDINE (5- β -D-ribofuranosyluracil) was the first naturally occurring C-nucleoside (C-glycosyl heterocycle) to be identified and studied.¹ The chemistry of pseudouridine (ψ) and other C-nucleosides as well as biochemical aspects have been reviewed extensively.² The enzymemediated synthesis of 2'-deoxypseudouridine-5'-phosphate was reported in $1972^{3,4}$ and this product was noted as an inhibitor of thymidylate synthetase.⁴ As suggested by the reaction conditions of Michelson and Cohn,⁵ cyclonucleoside formation is significantly more difficult with ψ than with uridine. Indeed, the easy $O-2 \rightarrow C-2'$ uridine cyclonucleoside closure employing diphenyl carbonate⁶ is unsuccessful (O-4 \rightarrow C-2') with ψ . The Mattocks-Moffatt⁷ reagent (α -acetoxyisobutyryl chloride) gives the O-2 \rightarrow C-2' cyclonucleoside and/or 2'-halogeno-2'-deoxy-ribo product with uridine.⁷ Our results with ψ show a different ultimate course and are at variance with recently published work.8-10

Treatment of ψ (1) with α -acetoxyisobutyryl chloride (neat) at 130 °C gave a mixture of chloro-sugar intermediates. Hydrogenolysis using tri-n-butyltin hydride and deblocking gave two major products whose ¹H n.m.r. and electron impact mass spectra were compatible with their assignment as 2'-deoxypseudouridine $(2'-d\psi)$ (2) and 3'-deoxypseudouridine $(3'-d\psi)$ (3). Two groups have very recently communicated syntheses of 2'-d ψ (2). Brown and his co-workers¹¹ employed condensation of a lithio pyrimidine and a protected aldehydo deoxy sugar to give 2'-d ψ (2) and its α -anomer, 5-(2-deoxy- α -D-erythropentofuranosyl)uracil (4). Watanabe et al. treated ψ (1) with α -acetoxyisobutyryl chloride in hot acetonitrile and obtained a mixture containing variously blocked cyclonucleosides and 2'-chloro-2'-deoxy derivatives.8-10 Hydrogenolysis of the mixture and deblocking resulted in isolation of two compounds which were assigned the β and α 2'deoxypseudouridine structures, (2) and (4), respectively.⁹ We have repeated the reaction of ψ and α -acetoxyisobutyryl chloride in hot acetonitrile (20 h) and have again obtained (2) and (3) (ca. 55:45) in ca. 90% combined vield after hydrogenolysis and deblocking.

Our data for the structure of β -2'-d ψ (2) are in agreement with the two previously noted studies.^{9,11} However, the structure of the second deoxynucleoside is now demonstrated to be 3'-d ψ (3) and we have observed no

evidence of any α -anomer. The ¹H n.m.r. spectrum [(CD₃)₂SO] of (3) has a doublet of doublets at δ 4.47 for the anomeric proton (H-1') with apparent coupling constants of *ca*. 1 and 2.4 Hz. The *ca*. 1 Hz splitting results from long-range coupling with H-6 of the uracil base. The $J_{1'-2'} = 2.4$ Hz coupling is consistent with values observed with 3'-deoxy- β -D-erythro-pentofuranosyl nucleosides derived from adenosine,¹² guanosine,[†] inosine,¹³ tubercidin,¹⁴ and formycin.¹⁴ The 3'-methylene protons of (3) give rise to a relatively narrow multiplet whereas the 2'-methylene protons of the α -anomer (4) were reported by Brown¹¹ to exhibit a distinctly separated pair of multiplet resonances. The H-4' resonance of the 3'-deoxy isomer is shifted downfield by *ca*. 0.4 p.p.m. at 100 MHz relative to that of the 2'-deoxy isomer as discussed previously.¹⁴



This trend is also apparent in reported data of Watanabe et al.^{9,10} As well, the u.v. c.d. spectra of ψ (1), 2'-d ψ (2), and (3) all have parallel negative long wavelength and positive shorter wavelength Cotton effect envelopes. The long wavelength transition amplitudes are closely similar for ψ (1) and (3), and somewhat reduced for 2'-d ψ (2). These Cotton effects are compatible with the β -3'-d ψ structure of (3), but would not be expected for an α anomer (whose c.d. spectrum should exhibit a long wavelength transition of reversed sign).¹⁵

Chemical degradation conclusively demonstrated the positional vs. stereo isomerism. The C-1'-O-4' bond of ψ was hydrogenolysed (rhodium catalyst¹) and the ribityl product was treated with sodium metaperiodate followed

by sodium borohydride. The resulting 5-(2-hydroxyethyl)uracil (5a) was examined by t.l.c. and was subjected to trimethylsilylation and mass spectrometric evaluation $[m/e \ 372 \cdot 1718, \text{ calc. for } (\mathbf{6a}) \ M^+ (C_{15}H_{32}N_2O_3Si_3) : 372 \cdot 1721].$ The identical reaction sequence was applied to $2' - d\psi$ (2) and compound (3). Mass spectra were compatible with formation of 5-(3-hydroxypropyl)uracil (5b) [m/e 386.1876], calc. for (6b) M^+ (C₁₆H₃₄N₂O₃Si₃): 386·1877] and 5-(2,4dihydroxybutyl)uracil (5c) [m/e 488.2384, calc. for (6c)]M⁺ (C₂₀H₄₄N₂O₄Si₄): 488.2378], respectively. T.l.c. indicated that (5b) and (5c) were exclusive products of the treatment of (2) and (3) with periodate-borohydride. When (5c) was again subjected to the oxidation-reduction sequence, no change was detected. This eliminates the possibility of prior incomplete periodate cleavage and affirms the 1,3-diol nature of (5c).

Thus, in contrast with the parallel reaction of uridine,7 ψ is converted by α -acetoxyisobutyryl chloride into a

mixture of 2'- and 3'-chlorodeoxynucleoside intermediates. Our full paper will describe reversible equilibrium mechanisms that rationalize the observed results in the published transformations of ψ based on our isolation of O-4 \rightarrow C-2' (cyclonucleoside) and C-2'-C-3' (epoxide) anhydro intermediates.

Biological testing inferences^{9,10} based on the assignment of the α -anomeric structure (4) to 3'-deoxypseudouridine (3) and the products derived therefrom via Fox and Watanabe's heterocyclic interconversions must now be reassessed.

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